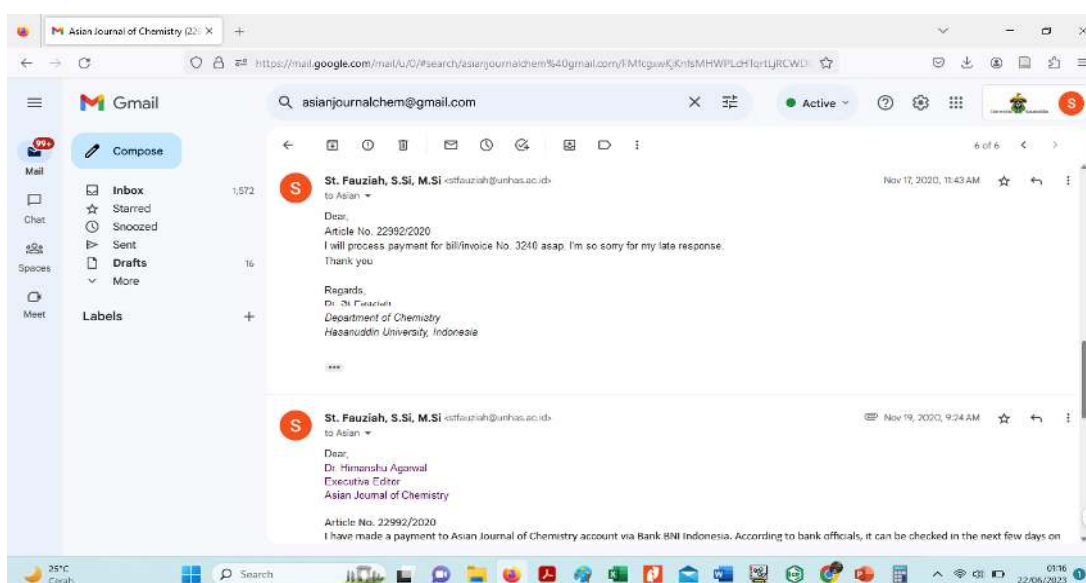
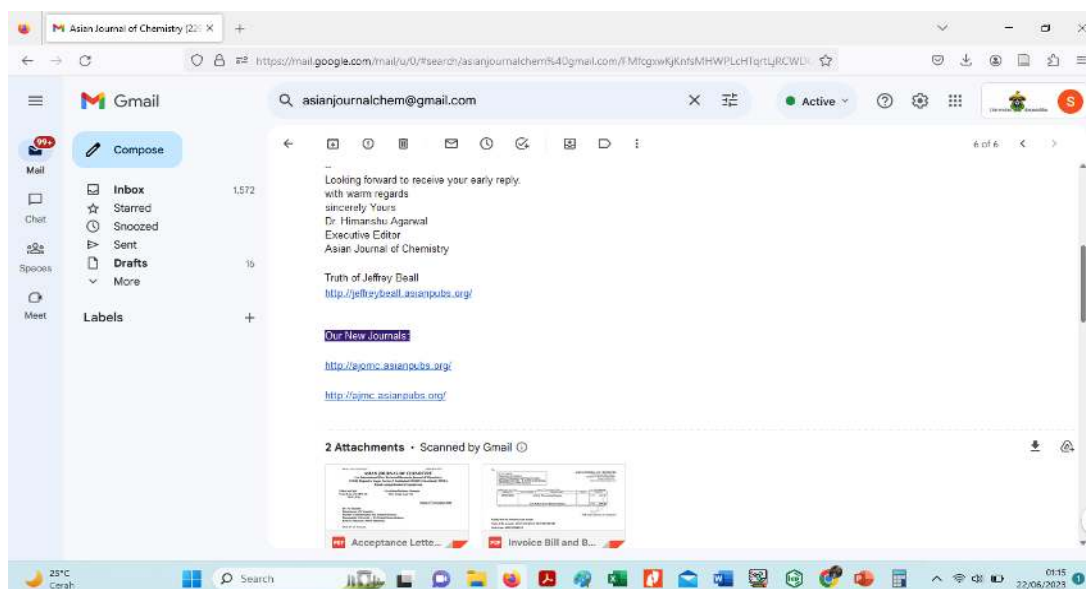
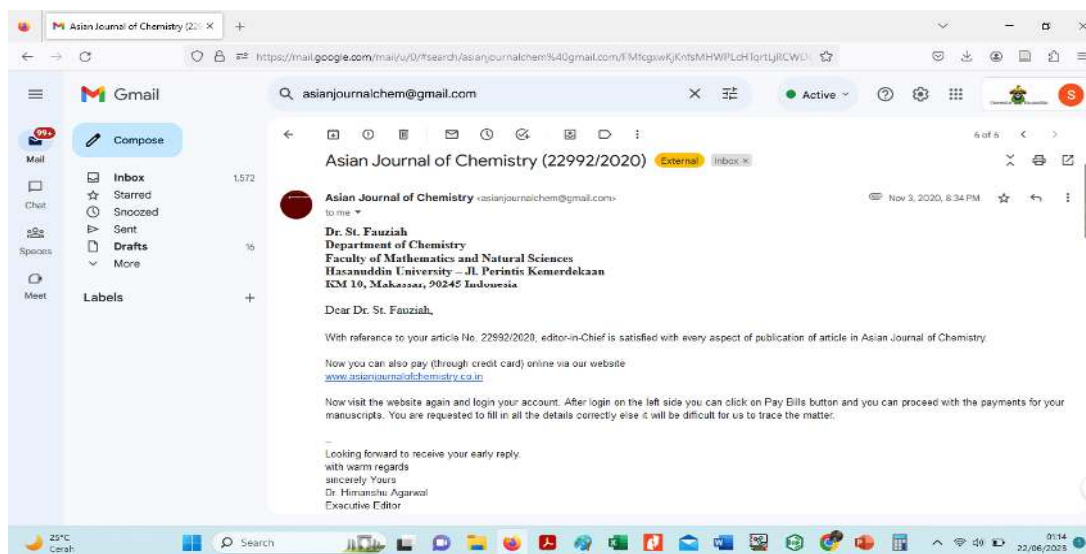


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ASIAN JOURNAL OF CHEMISTRY (AJC) Vol. 33 No. 4, pp. 785-788 (2021)

Judul Artikel "Synthesis of Molecularly Imprinted Polymers using Combination of Methacrylic Acid and Ethylene Glycol Dimethacrylate with Precipitation Polymerization Method"



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Synthesis of molecularly imprinted polymers using combination of methacrylic acid and ethyleneglycol dimethacrylate with precipitation polymerization method

ST FAUZIAH¹, F.A. SULLAH¹, N.H. SOEKAMTO¹, P. TABA¹, A. SAPAR²

¹ Department of Chemistry, Faculty of Mathematics and Natural Sciences, Hasanuddin University – Jl. Perintis Kemerdekaan, KM 10, Makassar, 90245 INDONESIA

² Department of Chemistry, Faculty of Mathematics and Natural Sciences, Tanjungpura University – Jl. Prof. Dr. Hadari Nawawi, Pontianak, 78124 INDONESIA

Best regards,
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² Department of Chemistry, Faculty of Mathematics and Natural Sciences, Tanjungpura University – Jl. Prof. Dr. Hadari Nawawi, Pontianak, 78124 INDONESIA

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Synthesis of molecularly imprinted polymers using combination of methacrylic acid and ethyleneglycol dimethacrylate with precipitation polymerization method

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¹ Department of Chemistry, Faculty of Mathematics and Natural Sciences, Hasanuddin University – Jl. Perintis Kemerdekaan, KM 10, Makassar, 90245 INDONESIA

² Department of Chemistry, Faculty of Mathematics and Natural Sciences, Tanjungpura University – Jl. Prof. Dr. Hadari Nawawi, Pontianak, 78124 INDONESIA

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Best regards,
Dr. St. Fauziah

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Dec 25, 2020, 11:01 PM

Dear,
Dr. Himanshu Agarwal
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I just want to ask about the progress of our article with the title "Synthesis of molecularly imprinted polymers using combination of methacrylic acid and ethylene glycol dimethacrylate with precipitation polymerization method". What volume and number will be published in the Asian Journal of Chemistry? Thankyou in advance.

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Best regards,
Dr. St. Fauziah
*Department of Chemistry, Faculty of Mathematics and Natural Sciences
Harammudin University, Makassar 21200/US14 90245
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Best regards,
Dr. St. Fauziah
Department of Chemistry, Faculty of Mathematics and Natural Sciences
Haramuddin University, Makassar, INDONESIA 90243
Email : stfauziah@unhas.ac.id

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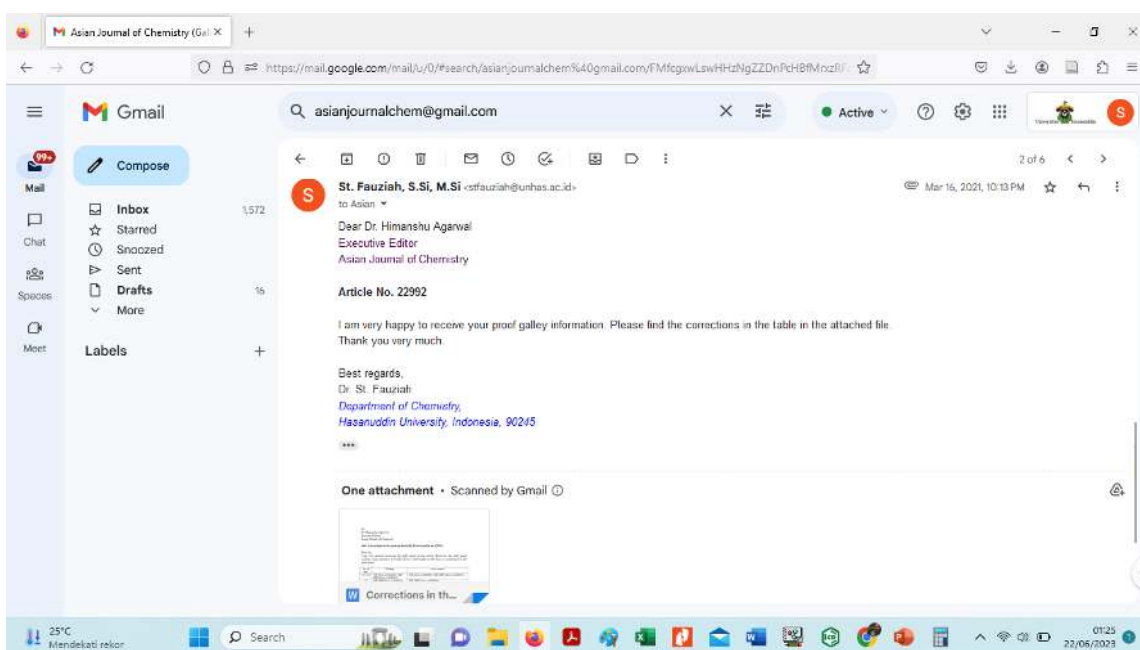
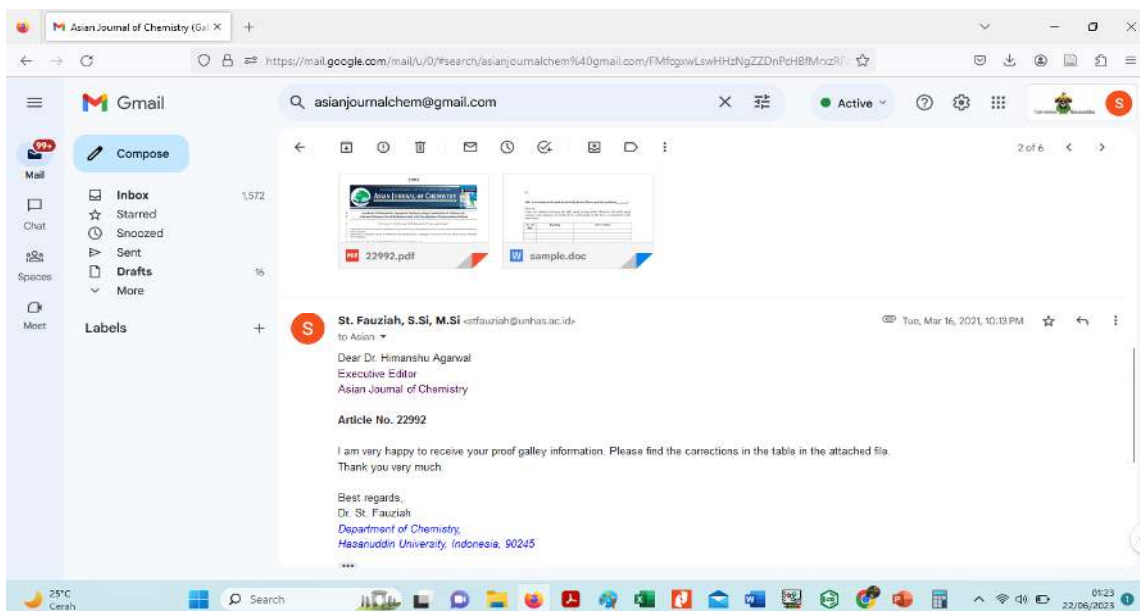
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Sub: Corrections to be made in the Gally Proof (article no 22992)

Dear Sir,

I am very pleased receiving the gally proof of our article. However, the gally proof contains some mistakes. I wonder if you could kindly rectify these as mentioned in the table below.

No. of line	Existing	to be written
125-126	NIP-MAA-co-EGDMA, MIP-DBP-MAA-co-EGDMA	NIP_MAA-co-EGDMA, MIP_DBP_MAA-co-EGDMA

127	MIP-DBP-MAA-co-EGDMA	MIP_DBP_MAA-co-EGDMA
129-130	MIP-DBP-MAA-co-EGDMA(BE) and MIP-DBP-MAA-co-EGDMA(AE)	MIP_DBP_MAA-co-EGDMA(BE) and MIP_DBP_MAA-co-EGDMA(AE)
132	MIP-DBP-MAA-co-EGDMA(AE)	MIP_DBP_MAA-co-EGDMA(AE)
Between 132-133	MIP-DBP-MAA-co-EGDMA(AE), (b) MIP-DBP-MAA-co-EGDMA(BE), (c) NIP_MAA-co-EGDMA and (d) MAA (in the title of Figure-1)	MIP_DBP_MAA-co-EGDMA(AE), (b) MIP_DBP_MAA-co-EGDMA(BE), (c) NIP_MAA-co-EGDMA and (d) MAA
Between 132-133	HE KEY FTIR DATA (in the title of Table-2)	THE FTIR DATA
133	MIP-DBP-MAA-co-EGDMA(BE)	MIP_DBP_MAA-co-EGDMA(BE)
136-137	NIP-MAA-co-EGDMA, MIP-DBP-MAA-co-EGDMA(BE) and MIP-DBP-MAA-co-	NIP_MAA-co-EGDMA, MIP_DBP_MAA-co-EGDMA(BE) and MIP_DBP_MAA-co-
142	NIP-MAA-co-EGDMA and MIP-DBP-MAA	NIP_MAA-co-EGDMA and MIP_DBP_MAA
155-156-157	NIP-MAAco-EGDMA, MIP-DBP-MAA-co-EGDMA(BE) and MIP-DBP-MAA-co-EGDMA(AE) and	NIP_MAA-co-EGDMA, MIP_DBP_MAA-co-EGDMA(BE) and MIP_DBP_MAA-co-EGDMA(AE) and
159	MIP-DBP-MAA-co-EGDMA(BE)	MIP_DBP_MAA-co-EGDMA(BE)
166	MIP-MAA-co-EGDMA	MIP_MAA-co-EGDMA
168-169	NIP-MAA-co-EGDMA, MIP-MAA-co-EGDMA(BE) and MIP-MAA-co-EGDMA(AE)	NIP_MAA-co-EGDMA, MIP_MAA-co-EGDMA(BE) and MIP-MAA-co-EGDMA(AE)
170	NIP-DBP-MAA	NIP_DBP_MAA
186	MIP-DBP-MAA-co-EGDMA	MIP_DBP_MAA-co-EGDMA
190	MIP-DEHP-MAA-co-EGDMA(AE)	MIP_DEHP_MAA-co-EGDMA(AE)
192	MIP-DBP-MAA-co-EGDMA	MIP_DBP_MAA-co-EGDMA
204	MIP-DBP-MAA-co-	MIP_DBP_MAA-co-
205	EGDMA(AE) to adsorb DBP is 60 min	EGDMA(AE) to adsorb DBP is 60 min and the amount of adsorbed DBP is 1.1746 mg/g.
Between 205-206	MIP-DBP-MAAco-EGDMA(AE) (in the title of Figure-5)	MIP_DBP_MAA-co-EGDMA(AE)
206	Effect of concentration on MIP adsorption: Missing	Effect of concentration on Adsorption of DBP by MIP Figure 6 shows the relationship between the concentration and the amount of DBP adsorbed by MIP, namely the higher the initial concentration of the DBP standard solution, the more DBP can be adsorbed by MIP_DBP_MAA-co-EGDMA (AE). However, if the maximum limit has been reached in the adsorption equilibrium, the

adsorption capacity of MIP_DBP_MAA-co-EGDMA (TE) will tend to be constant even though the concentration is increased. The effect of concentration on the adsorption ability of DBP is shown in Figure 6.

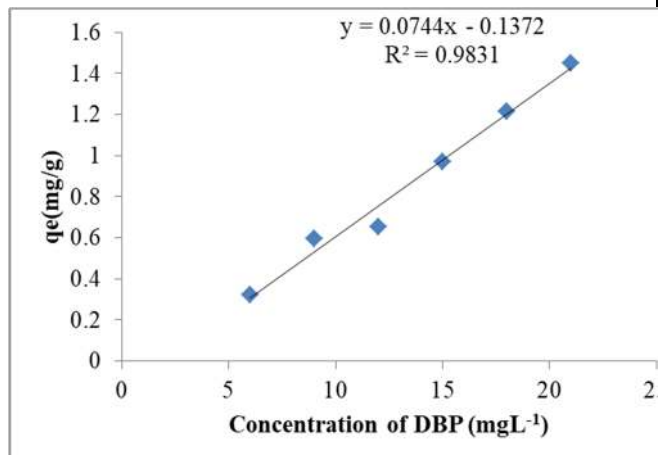


Figure 6. The effect of concentration on the adsorption ability of MIP_DBP_MAA-co-EGDMA_(AE) to DBP

Below are the 4 author's ORCID IDs :

1. St Fauziah <https://orcid.org/0000-0002-0599-4011>
2. Nunuk Hariani Soekamto <https://orcid.org/0000-0001-8281-1752>
3. Paulina Taba <https://orcid.org/0000-0001-7327-5505>
4. Ajuk Sapar <https://orcid.org/0000-0002-7278-6159>
5. Fausan Akram Sullahi is a college student and he do not has Orcid ID

Thank you very much.

Best regards,

Dr. St. Fauziah
Department of Chemistry,
Hasanuddin University, Indonesia